



TITLE:

# On the Molecular Configurations of $\gamma$ -BHC, $\delta$ -and $\epsilon$ -1,1,2,3,4,5,6-Heptachlorocyclohexane

AUTHOR(S):

Oiwa, Toshihiko; Yamada, Ryoichi; Ohno, Minoru

---

CITATION:

Oiwa, Toshihiko ...[et al]. On the Molecular Configurations of  $\gamma$ -BHC,  $\delta$ -and  $\epsilon$ -1,1,2,3,4,5,6-Heptachlorocyclohexane. 京都大学化学研究所報告 1951, 24: 73-73

ISSUE DATE:

1951-03-30

URL:

<http://hdl.handle.net/2433/74237>

RIGHT:

### 13. On the Molecular Configurations of $\gamma$ -BHC, $\delta$ - and $\epsilon$ -1, 1, 2, 3, 4, 5, 6-Heptachlorocyclohexane

Toshihiko Oiwa, Ryoichi Yamada and Minoru Ohno

(Takei Laboratory)

It has been already reported that by the present authors (Botyu-Kagaku, **15**, 86 (1950)<sup>(\*)</sup>) a new isomer of 1, 1, 2, 3, 4, 5, 6-heptachlorocyclohexane (mp. 55-55.5°;  $\epsilon$ -hepta) was obtained from the chlorination product of  $\gamma$ -BHC. At this time  $\delta$ -1, 1, 2, 3, 4, 5, 6-heptachlorocyclohexane (mp. 139-140°;  $\delta$ -hepta) and  $\epsilon$ -hepta were isolated chlorination product of  $\alpha$ -BHC with  $\gamma$ -1, 1, 2, 3, 4, 5, 6-heptachlorocyclohexane (mp. from the 85-86°;  $\gamma$ -hepta) and o-1, 1, 2, 2, 3, 4, 5, 6-octachlorocyclohexane by partition chromatography.

The molecular configuration of  $\alpha$ -BHC has been already determined as is shown in Table (Botyu-Kagaku, **15**, 32(1950)). The possible isomers of 1, 1, 2, 3, 4, 5, 6-heptachlorocyclohexane (hepta), which can be derived from this, are II, III, and IV. Since II has been assigned to be the molecular configuration of  $\gamma$ -hepta<sup>(\*)</sup>, one of the two forms left is of  $\delta$ -hepta, and the other of  $\epsilon$ -hepta. Now, taking into account the fact that the forms III and IV can be also derived from the forms VII and V of 16 possible isomers of BHC respectively, and the experimental results that  $\epsilon$ -hepta is also produced by chlorination of  $\gamma$ -BHC, it must be said that one of the two, V or VII, is the molecular configuration of  $\gamma$ -BHC. As has been pointed out by Y. Morino et al. (Botyu-Kagaku, **15**, 181 (1950)), the calculated values of dipole moments of the two forms are 3.19-2.93 D (V) and 1.88 D (VII), and the experimental value for  $\gamma$ -BHC is 2.80 D. Consequently, V should be the molecular configuration of  $\gamma$ -BHC. The isomers of hepta which can be derived from V are IV, IV', VI and VI', but among them only IV can be obtained by the chlorination of both  $\alpha$ - and  $\gamma$ -BHC. Therefore, the conclusion is that IV is  $\epsilon$ -hepta and III is  $\delta$ -hepta.

Table : The Chlorine Configurations \*\*

I. $\alpha$ -BHC .....	p, p, e, e, e, e	V. $\gamma$ -BHC.....	p, p, e, e, e, e
II. $\gamma$ -hepta .....	p, $\widehat{pe}$ , e, e, e, e	IV.' .....	p, p, p, $\widehat{pe}$ , e, e
III. $\delta$ -hepta .....	p, p, e, $\widehat{pe}$ , e, e	VI. ....	p, $\widehat{pe}$ , p, e, e, e
IV. $\epsilon$ -hepta .....	p, p, $\widehat{pe}$ , e, e, e	VI.' .....	p, p, p, e, $\widehat{pe}$ , e
		VII. ....	p, p, e, p, e, e

\*\* This is shown by simple notation of p (polar) and e (equatorial) proposed by C. W. Bekett et al. (J. Am. Chem. Soc., **69**, 2488 (1947)).